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Gizmo illustrates how the molecules  
of gas move at different temperatures.  
In this Gizmo, temperature is  
measured on the Kelvin scale, which  
measures temperature from absolute  
zero, the coldest possible temperature  
(-273.15 °C).

Student Exploration: Temperature and

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## Particle Motion

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Particle Motion Question: How is the temperature of a gas related to the motion of gas molecules? 1. Observe: Move the Temperature slider back and forth. Focus on the particle motion at left. What do you notice? The colder it gets the slower they go the hotter it gets the faster they will go.

Copy of R Temperature and Particle Motion.docx - Student ...

Gizmo Warm-up The Temperature and Particle Motion Gizmo illustrates how the molecules of gas move at different temperatures. In this Gizmo, temperature is measured on the Kelvin scale, which measures temperature from absolute zero, the coldest possible temperature ( $-273.15\text{ }^{\circ}\text{C}$ ).

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Student Exploration- Temperature and Particle Motion ...

Name: Anaya Tei Date: October 23,2020 Student Exploration:

Temperature and Particle Motion

Vocabulary: absolute zero, Kelvin scale, kinetic energy, Maxwell-

Boltzmann distribution, molar mass, molecule, temperature, universal gas

constant Prior Knowledge Questions (Do these BEFORE using the Gizmo.)

1. Why is hot air hot? Hot air is hot because the sun is radiating hot oxygen 2.

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The Temperature and Particle Motion

Gizmo illustrates how the molecules of gas move at different temperatures.

In this Gizmo, temperature is

measured on the Kelvin scale, which

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measures temperature from absolute zero, the coldest possible temperature (-273.15 °C).

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Student Exploration: Temperature and Particle Motion Vocabulary: absolute zero, Kelvin scale, kinetic energy, Maxwell-Boltzmann distribution, molar mass, molecule, temperature, universal gas constant Prior Knowledge Questions (Do these BEFORE using the Gizmo.) 1.

Temperature\_and\_Particle\_Motion\_Gizmo.docx - Name Date ...

Student Exploration: Temperature and Particle Motion 4 Prior Knowledge Questions (Do these BEFORE using the Gizmo.) 1. Why is hot air hot? Hot

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air rises because when you heat air (or any other gas for that matter), it expands. When the air expands, it becomes less dense than the air around it.

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Student Exploration: Temperature and Particle Motion. Vocabulary: absolute zero, Kelvin scale, kinetic energy, Maxwell-Boltzmann distribution, molar mass, molecule, temperature, universal gas constant. Prior Knowledge Q. uestions (Do these BEFORE using the Gizmo.) Why is hot air hot? \_\_\_\_\_

Temperature and Particle Motion  
In the Temperature and Particle Motion Gizmo, students explore how the temperature and molecular weight



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of a gas relates to the distribution of particle velocities. The Gizmo includes a simulation that shows how particles in a gas collide and how momentum and kinetic energy are transferred between particles.

Gizmo of the Week: Temperature and Particle Motion ...

Temperature and Particle Motion

Observe the movement of particles of an ideal gas at a variety of temperatures. A histogram showing the Maxwell-Boltzmann velocity distribution is shown, and the most probable velocity, mean velocity, and root mean square velocity can be calculated. Molecules of different gases can be compared.

Temperature and Particle Motion  
Gizmo : Lesson Info ...

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Answers  
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## Answers

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Gizmo illustrates how the molecules

of gas move at different temperatures.

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In this Gizmo, temperature is measured on the Kelvin scale, which measures temperature from absolute zero, the coldest possible temperature (-273.15 °C).

Solubility And Temperature Gizmo  
Answer Key Activity A

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